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# Effect of Titania Nano-additives on Fine and Ultrafine Carbonaceous Emissions during Flame Combustion of Diesel

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# ABSTRACT

The severe impacts of emissions from combustion necessitate the need for advanced mitigation techniques. This study focuses on experimentally investigating the change in particulate matter (PM) emissions during the flame combustion of diesel blended with varying quantities of titania (TiO<sub>2</sub>) nano-additives. The initial observations using a Scanning Mobility Particle Sizer showed a reduction in the total number concentration (TNC) of PM emissions for TiO<sub>2</sub>-nanofuel samples compared to diesel. However, detailed investigations revealed the enhancement in the TNC of ultra-fine particles (UFPs) with mobility diameters less than 100 nm. This indicates the possibility of the emission of nano-additives during flame combustion, which enhances the number concentration of UFPs. The evolution of  $TiO_2$  nanoparticles is validated by performing the elemental composition analysis using energy-dispersive X-ray spectroscopy after sampling the PM emissions. A detailed experimental study also revealed the significance of the size and stability of the dispersed nanoparticles in the overall emissions. Ball milling (BM) was used for the size reduction of dispersed nanoparticles to enhance the dispersibility of the nano-additives. BM, when combined with bathsonication (BS), resulted in the highest reduction in the TNC (37.70% for Ti100 BM-BS, 48.46% for Ti150 BM-BS, and 53.27% for Ti200 BM-BS), highlighting the importance of size of the dispersed nanoparticles. The detailed analysis of UFPs showed an increase in the TNC of particulates in sub-23 nm (22.92% for Ti100 BS and 39.16% for Ti100 BM) and super ultra-fine (96.46% for Ti100 BS, 100.83% for Ti100 BM, and 16.73% for Ti100 BM-BS) regions for nanofuel samples in contrast to neat diesel.

Keywords: Ball milling, Combustion aerosols, Nano-additives, Particle size distribution, Ultra-fine particles

# **1 INTRODUCTION**

Particulate and gaseous emissions from various combustion sources have far-reaching implications on indoor and outdoor air quality (Gaffney and Marley, 2009; Vardoulakis *et al.*, 2020). Pollutants generated during the combustion of diesel fuel typically comprise particulate matter (PM), carbon monoxide (CO), nitrogen oxides (NO<sub>x</sub>), volatile organic compounds (VOCs), etc. Fine and ultra-fine particulates have an adverse effect on living beings and the environment (Maciejczyk *et al.*, 2021; Sher, 1998). Chronic exposure to PM emissions is associated with severe respiratory conditions like asthma, bronchitis (Walsh, 2014; WHO, 2018). Ultra-fine particles (UFPs) are even more severe in affecting human health as they can deeply infiltrate the respiratory system and penetrate the bloodstream, exacerbating other health issues such as cardiovascular diseases, gastrointestinal



diseases, and neurological disorders (Cassee et al., 2019; HEI, 2013; Sonwani et al., 2021).

The transportation sector is a prominent source of particulate emissions due to its reliability on conventional fuels. Despite the prevailing options of alternative energy sources, diesel persists as the predominant fuel choice within the transportation sector. The compression ignition (CI) technology uses diffusive flame combustion where the imperfect mixing of fuel with air (oxidizer) leads to the formation of high concentrations of carbonaceous aerosols (Liu et al., 2020). Over the last two decades, nanoparticles as additives in liquid fuels have been used, which improves the combustion characteristics of fuel and simultaneously reduces harmful emissions (Ben Said et al., 2022; Tomar and Kumar, 2019). Nano-additives have a high specific surface area, leading to improvement in the heat transfer characteristics of the fuel. Additionally, nano-additives act as catalyst and enhance the combustion process, promoting better oxidation of fuel and resulting in the reduction of gaseous and PM emissions. The droplet combustion studies have also shown the phenomenon of more intense and frequent micro-explosions in the case of nanofuel compared to neat diesel (Mehrizi et al., 2022). The micro-explosion of primary atomized droplets leads to secondary atomization that aids in the better mixing of fuel with air. This helps in better combustion of diesel fuel, leading to a reduction in unburnt hydrocarbon, carbon monoxide, soot, and other particulate emissions. Therefore, nano-additives, when blended with diesel, improve the thermo-physical properties and combustion characteristics of the fuel, also resulting in the reduction of PM and gaseous emissions (Dobrzyńska et al., 2022; Sezer, 2019). The combustion process in the Cl engine includes the spraying of fuel into the combustion chamber, mixing of atomized droplets with air, evaporation of liquid droplets into vapor, burning of air-fuel mixture, and the diffusive flame ignited around the sprayed droplets helps the chemical reaction to proceed further. Thus, the fundamental study of a diffusive flame consists of the spray injection characteristics, droplet evaporation characteristics, flame combustion characteristics, etc. Investigations have been done to study the benefits of adding nanoparticles to diesel and biodiesel blends on the CI engine characteristics (Basha et al., 2022; Gad et al., 2023). The experimental analysis of the droplet evaporation characteristics validates the phenomenon of micro-explosions during nanofuel combustion (Mehrizi et al., 2022). Consequently, there is an improvement in the combustion phenomenon, leading to a reduction in PM and gaseous emissions. The diffusive flame characteristics during combustion are fundamentally studied through strand experiments. However, the elementary study of PM emissions for liquid fuel diffusion flame still needs detailed investigations. Additionally, the effect of blending nano-additives to diesel on PM emissions reduction and changing the number concentration of UFPs is guite essential.

Although various researchers have investigated the benefits of adding nanoparticles to diesel and diesel-biodiesel blends, very few studies have investigated the role of the primary particle size in the improvement of combustion and emission characteristics of the fuel (Ağbulut, 2022; Chinnasamy *et al.*, 2019; Dinesha *et al.*, 2021). However, none of the previous work has studied the actual size of particles/aggregates dispersed in diesel/biodiesel and its importance in the reduction of emissions. The aggregation of nanoparticles dispersed in diesel enhances the sedimentation process, and the benefits of adding nanoparticles get nullified over time. Therefore, the stability in terms of dispersibility of nanoparticles is an essential aspect that considers the shelf life of nanofuel. The particle size distribution (PSD) of the dispersed nano-additives aids in understanding the significance of the size of nanoparticles in the reduction of fine and ultra-fine particulate emissions. The reduction in the size of the dispersed nanoparticles using mechanical methods leads to improvement in the shelf life of nanofuel suspension.

The fundamental study of the flame combustion of fuel blended with nano-additives provides better insights for the reduction in particulate matter emissions. The experimental investigations showed the suppression of soot emissions from a propane diffusion flame with the addition of metal additives (Cotton *et al.*, 1971). There is the production of hydroxyl radicals, which rapidly remove soot or soot precursors. The results also showed that the soot reduction mechanism changes with the type of metal. Similarly, the sooting tendency increases for laminar pre-mixed ethylene-oxygen-argon flames with the addition of ferrocene particles (Hirasawa *et al.*, 2004). Both experimental and computational studies support the enhancement in soot formation as ferrocene particles provide a surface to initiate soot structure growth. Further, experiments were conducted to measure the soot emission from turbulent methane and propane diffusion flames



using laser-induced incandescence (LII) (Canteenwalla et al., 2007). The results show that the amount of soot formation depends upon fuel composition and the fuel flow rate through the burner. In addition, the research was carried out to analyze the effect of the addition of ethanol to gasoline on the soot formation in a co-flow diffusion flame (Matti Maricq, 2012). The study included the comparison of soot emissions with flame height, and the results showed the reduction in volatile organic compounds (VOCs) and soot primary particle size with the percentage increment of ethanol blended. The burning rate study of liquid monopropellants with and without additives was carried out using a guartz-lined cavity and steel tube (McCown et al., 2014). The results for strand burner tube experiments showed an increase in burning rates by 60–70% with the addition of 1% silica and 5% aluminum to neat nitromethane. Furthermore, the experiments were performed to understand the growth of soot particles and polycyclic aromatic hydrocarbons (PAHs) at atmospheric pressure for fuel-rich propane/oxygen-diffusion flame with the addition of 0.1% and 0.2% ferrocene metal additives (Hu et al., 2016). The results show that ferrocene reacts with O and OH radicals, aids in lowering the burner flame temperature, and restrains the growth of PAHs into larger aggregates. Additionally, there is better oxidation of soot particles at lower temperatures when fuel is blended with ferrocene additives. Similarly, soot particle size distribution for laminar pre-mixed burner-stabilized ethylene flames has been studied (Zhou et al., 2023). The experimental and computational study shows enhanced particle coagulation and the growth in soot particle size with the addition of ferrocene additives.

The benefits of adding nanoparticles to reduce PM emissions are quite evident. However, the experimental studies for PM emissions, including UFPs, during diesel flame combustion need detailed investigations. The effect of the addition of nano-additives on the emissions of fine and ultra-fine particulate matter is still not well understood. The measurement of the number concentration of particulate emissions helps us to understand the physical mechanism behind the corresponding increase or decrease in PM emissions for the distinct size range. Although several studies across the world have looked into the overall effect of nano-additives in terms of PM concentration, combustion characteristics, etc., very few studies have attempted to understand the evolution of the nano-additives post-combustion and their impact on UFP emissions (Kegl *et al.*, 2021). Thus, the present study includes the qualitative and quantitative analysis of the PM emitted during the flame combustion of diesel with and without the addition of TiO<sub>2</sub> nano-additives, with a special focus on submicron-sized carbonaceous particles.

The strand experiments are significant in carrying out the fundamental studies for a better understanding of the combustion process. The experiments of PM emissions measurement were conducted in a controlled environment that allows the easy reproducibility of the data. The stability of nanofuel is also an essential aspect that needs a detailed study to effectively utilize the benefits of nano-additives for a long time. Consequently, this study focuses on the use of ball milling in altering the particle size distribution (PSD) of the TiO<sub>2</sub> nanoparticles dispersed in diesel. The dispersibility of nanoparticles helps to understand the shelf life of nanofuel. In this study, the experimental measurements of the PM emissions during the flame combustion of diesel were carried out using a Scanning Mobility Particle Sizer (SMPS) to understand the effect of nanoadditives. The significance of the size and morphology of TiO<sub>2</sub> nanoparticles in the reduction of total PM emissions is investigated. Presently, the focus is on the impact of nanoparticles on the ultra-fine particulate emissions and the change in the number concentration of PM emissions in sub-23 nm (< 23 nm), super-ultra-fine (23–50 nm), and ultra-fine (50–100 nm) regions. The characterization of the PM emissions using field emission scanning electron microscopy (FESEM) and energy-dispersive X-ray spectroscopy (EDS) to investigate the evolution of nano-additives during flame combustion is also reported.

### **2 MATERIALS AND EXPERIMENTAL METHODS**

The strand experiments were performed with diesel fuel (purchased from a local BPCL petroleum station). Titania (TiO<sub>2</sub>) nanoparticles (average particle size, APS  $\leq$  25 nm) were purchased from Sigma-Aldrich Chemicals Pvt. Ltd., and their specifications are summarized in Table S1. The field emission scanning electron microscopy (FESEM) images and energy-dispersive X-ray spectroscopy

(EDS) analysis of titania nano-additives are presented in Fig. S1 and Table S2, respectively.  $TiO_2$  nano-additives were blended into neat diesel in the quantity of 100 ppm, 150 ppm, and 200 ppm to prepare the nanofuel suspensions.  $TiO_2$ , being a metal oxide, acts as a catalyst and promotes better oxidation of fuel, leading to improvement in the combustion process. Therefore,  $TiO_2$  nano-additives aid in the reduction of soot and other PM emissions. Since  $TiO_2$  is known to be biocompatible and is safe for human health when exposed under standard conditions, its existence in PM emissions post-combustion might not present a health risk (Rashid *et al.*, 2021).

#### 2.1 Nanofuel Sample Preparation

The nanofuel samples for experimental investigations are prepared by adding TiO<sub>2</sub> nanoparticles to diesel. The nanofuel suspension was hand-stirred to prepare a homogeneous mixture. When TiO<sub>2</sub> nanoparticles were blended with diesel, the particles tend to form aggregated structures, and the average size of the dispersed TiO<sub>2</sub> nanoparticles can be much larger than the APS mentioned in Table S1. Thus, Span80 (sorbitan monooleate) surfactant was added to diesel to hinder the aggregation of the dispersed nanoparticles. The addition of surfactant to diesel should not alter the properties of diesel significantly; therefore, the amount of surfactant added was twice that of the nano-additives. TiO<sub>2</sub> nano-additives were dispersed in diesel fuel and Span80 mixture in ppm (or mass percentage ratio). After hand stirring for 15 minutes, the nanofuel sample was ultra-sonicated for 60 minutes with the help of the Branson CPX3800H bath sonicator (BS), purchased from Emerson Electric.

The blending of nano-additives to diesel is beneficial due to the high specific surface area of the nanoparticles added. Consequently, the morphology of TiO<sub>2</sub> nanoparticles plays a vital role during the combustion of TiO<sub>2</sub>-laden diesel nanofuel (Jeon *et al.*, 2015). Considering this, the size modification of the dispersed nano-additives was done using a mini-planetary twin-bowl holder ball mill MBM-07, purchased from InSmart Systems. The ball milling (BM) was done for 32 minutes, and the other input parameters used for BM the wet mixture of TiO<sub>2</sub> nano-additives, Span80, and diesel are given in Table S3. Once the BM operation was completed, the wet mixture was poured back into the bottle, hand-stirred for 15 minutes, and ultra-soniacted for 60 minutes.

For the stability analysis, particle size distribution (PSD) of the dispersed  $TiO_2$  nano-additives was measured by the Dynamic Light Scattering (DLS) technique using a Zetasizer ZS90, purchased from Malvern Panalyticals. DLS measures the concentration of dispersed particles as a function of their hydrodynamic diameter ( $D_h$ ). Due to the effect of rotational diffusion, PSD analysis was in terms of apparent hydrodynamic diameters and not true hydrodynamic diameters. The following naming of the test fuel samples is used throughout the manuscript:

- 1) Diesel (neat diesel)
- 2) Ti100 no-Sp BS (100 ppm TiO<sub>2</sub> + diesel, bath-sonicated)
- 3) Ti100 BS (100 ppm TiO<sub>2</sub> + 0.02% wt. Span80 + diesel, bath-sonicated)
- 4) Ti100 BM (100 ppm TiO<sub>2</sub> + 0.02% wt. Span80 + diesel, ball-milled)
- 5) Ti100 BM-BS (100 ppm TiO<sub>2</sub> + 0.02% wt. Span80 + diesel, ball-milled, bath-sonicated)
- 6) Ti150 BS (150 ppm TiO<sub>2</sub> + 0.03% wt. Span80 + diesel, bath-sonicated)
- 7) Ti150 BM (150 ppm TiO<sub>2</sub> + 0.03% wt. Span80 + diesel, ball-milled)
- 8) Ti150 BM-BS (150 ppm TiO<sub>2</sub> + 0.03% wt. Span80 + diesel, ball-milled, bath-sonicated)
- 9) Ti200 BS (200 ppm TiO<sub>2</sub> + 0.04% wt. Span80 + diesel, bath-sonicated)
- 10) Ti200 BM (200 ppm TiO<sub>2</sub> + 0.04% wt. Span80 + diesel, ball-milled)
- 11) Ti200 BM-BS (200 ppm TiO<sub>2</sub> + 0.04% wt. Span80 + diesel, ball-milled, bath-sonicated)

The effect of the morphology of dispersed nano-additives on flame combustion is examined by comparing the fine and ultra-fine particulate matter emissions in the case of bath-sonicated and ball-milled nanofuel samples with diesel.

#### **2.2 Properties of the Test Fuel Samples**

The important thermo-physical properties of the test fuel samples are density ( $\rho$ ) and calorific value (CV). The mass of TiO<sub>2</sub> nano-additives, surfactant, diesel fuel, etc., was measured using an electronic weighing balance (Radwag AS 220.R2, maximum load = 220 g). The density of diesel and nanofuel samples was calculated by measuring the mass of the known volume of fuel. The CV of

test fuel samples was measured using a mLabs bomb calorimeter purchased from Labtronics. After the measurement of the properties of test fuel samples, the experiments were carried out for PM emissions measurement during flame combustion of TiO<sub>2</sub>-laden diesel fuel.

#### 2.3 Experimental Setup

The PM emissions study through strand experiments aids in understanding the effect of nanoadditives on fine and ultra-fine particulates in a controlled way. The morphology of nanoparticles is an important factor that governs the improvement in thermo-physical properties and combustion characteristics of the fuel. Therefore, the size and structure of  $TiO_2$  nanoparticles play a crucial role in the suppression of PM emissions during the flame combustion of diesel. To facilitate that, Span80 surfactant and mechanical methods (ball milling and bath sonication) were used for nanofuel preparation that helped to hinder the aggregation mechanism and size reduction of dispersed nano-additives, respectively.

The present study for strand experiments of flame combustion of diesel with the addition of TiO<sub>2</sub> nano-additives was carried out in a quiescent ambience at a standard temperature and atmospheric pressure (298.15 K, 1 atm.). The PM, including UFPs, emissions were measured with the help of a Scanning Mobility Particle Sizer (SMPS) purchased from TSI Incorporated. SMPS measures the PSD based on mobility diameter (D<sub>m</sub>) and quantifies the variation in normalized number concentration with D<sub>m</sub> of the particles present in the aerosol (PM emissions) inflow. Additionally, SMPS provides the total number concentration (TNC) of PM emissions for the measured size range. For the measurement of PM emissions, the hollow non-conducting tube was fixed at a height of 1 foot, just above the rim of the test tube. The tube was connected to the input of the Classifier through a silica gel for the intake of aerosol. Silica gel was used to remove the water vapor content. The aerosol flows from the Classifier to the Differential Mobility Analyzer (DMA) and then to the Condensation Particle Counter (CPC). A steady sheath-air flow rate (SFR) was maintained inside DMA to ensure a controlled and stable flow environment. The schematic diagram of the experimental set-up is shown in Fig. 1. The collection of PM emissions for FESEM characterization and EDS analysis was done at the same height and location where the SMPS measurements were done.

After preparation, the nanofuel samples were poured into a 3 mL test tube hung vertically using a clamp holder. The test tube was filled with the fuel sample completely up to the rim in each case to minimize any experimental error. The fuel was ignited using a pilot flame with the help



**Fig. 1.** Schematic diagram of the experimental set-up for PM emissions measurement during flame combustion (strand experiments).



of a gas lighter, which is extinguished as soon as flame combustion takes place above the test-tube rim. The representation of flame for diesel and a few nanofuel samples is shown in Fig. S5.

#### **2.4 Uncertainty Analysis**

All the experiments for the measurement of properties, DLS analysis for nanofuel stability, and PM emissions measurement were carried out three times each for better statistical analysis. The study of uncertainty consists of the measurement of standard deviation ( $\sigma$ ) for each set of experiments and indicated in the respective plots as error bars. The relative standard deviation ( $\sigma_{rel}$ ) aids in understanding the deviation in measured values from the mean value in an effective manner. Table S4 shows the uncertainty analysis in terms of  $\sigma_{rel}$  for various experimental measurements. The uncertainty analysis during DLS measurements is shown in Fig. 2 (Results section 3.1) for a better understanding of the average results. The standard deviation in the measurement of the total number concentration of PM emissions is corroborated in the respective plot (Fig. 5(a)). Similarly, the standard deviation in the measurement of the normalized number concentration of PM emissions at distinct mobility diameter (D<sub>m</sub>) is presented separately in Fig. S4 to avoid confusion in the graphical comparison of the number concentration for different test fuel samples.

## **3 RESULTS AND DISCUSSION**

The stability study of nanofuel suspension in terms of dispersibility of the  $TiO_2$  nanoparticles consists of the PSD plots with the intensity of light scattered. The PM emissions investigation during flame combustion includes the comparison of the number concentration with distinct mobility diameter ( $D_m$ ) in the case of nanofuel samples with diesel. The TNC of UFPs and total PM emissions are analyzed to effectively understand the impact of  $TiO_2$  nano-additives.

#### **3.1 Nanofuel Stability**

The gravitational settling of TiO<sub>2</sub> nano-additives depends on their shape, size, and density. Nanoparticles, once dispersed in diesel, tend to aggregate and form large aggregated structures due to various collision mechanisms, including Brownian motion. Thus, the average size of the dispersed particles/aggregated structures will increase, resulting in an increased sedimentation rate of dispersed nanoparticles even though the drag force is also affected. This accelerates the settling of TiO<sub>2</sub> nanoparticles with time, which nullifies the benefits of using nano-additives (Xu, 2018; Zhang, 2014). Therefore, Span80 was used to provide steric stabilization to TiO<sub>2</sub> nanoparticles.

To study the effect of size and morphology on the stability of nanofuel suspension, nanofuel suspension was ball-milled to reduce the size of the dispersed particles. Considering the advantages of mechanical methods in nanoparticle dispersibility, BM and BS were used to reduce the size of the dispersed particles. The PSD analysis presented in Fig. S2 clearly shows the reduction in the



**Fig. 2.** DLS measurement of the nanofuel samples for 15 days: average apparent hydrodynamic diameter  $(D_{h,avg})$  of the TiO<sub>2</sub> nanoparticles dispersed in diesel.

size of the dispersed particles. The long-term stability of nanofuel samples (Ti100 BS, Ti100 BM, Ti100 BM-BS) is studied for 15 days using DLS, and the intensity plots are compared in Fig. S3. The PSD of dispersed TiO<sub>2</sub> nanoparticles in Ti100 BM-BS nanofuel shows the presence of a significant concentration of particles in the submicron- and fine-size range, even after 15 days of nanofuel sample preparation. Fig. 2 shows the average apparent hydrodynamic diameter ( $D_{h,avg}$ ) of the nanoparticles measured up to 15 days. The nanofuel sample, which is ball-milled and bath-sonicated (Ti100 BM-BS), shows the minimum  $D_{h,avg}$  and the maximum dispersibility among all the nanofuel samples, clearly demonstrating the positive effects of size reduction of the dispersed particles (Jain *et al.*, 2023).

#### **3.2 Properties of the Test Fuel Samples**

The comparison of thermo-physical properties ( $\rho$ , CV) of nanofuel with diesel is presented in Table 1. The density of TiO<sub>2</sub> nanoparticles is much higher as compared to neat diesel; therefore, there is a slight increment in the density of nanofuel samples. The blending of nanoparticles into diesel improves the combustion process and heat transfer characteristics of fuel. However, there will be minor effects of the Span80 surfactant on the CV of the fuel. Additionally, the measurement of the CV of different test fuel samples has some experimental errors. Thus, the change in the CV of the fuel is insignificant with the addition of TiO<sub>2</sub> nanoparticles, supporting prior investigations (Örs *et al.*, 2018; Soudagar *et al.*, 2018).

#### **3.3 FESEM Characterization and EDS Analysis**

The characterization of PM emitted during the flame combustion of  $TiO_2$ -laden nanofuel samples was done using FESEM, as shown in Fig. 3. The EDS analysis shown in Table 2 confirms the evolution of  $TiO_2$  nanoparticles during the flame combustion of nanofuel samples. The evolution of  $TiO_2$  nano-additives affects the number concentration of PM emissions, which is discussed in detail in the further section (Results section 3.4).

#### **3.4 PM Emissions Measurement and Analysis**

The PM emissions, including UFPs, were measured using SMPS. Initially, the PM present in ambient air was measured to set the baseline to investigate the effect of nano-additives in PM emissions reduction effectively. The total number concentration (TNC) for ambient air was

Properties	Diesel	Ti100 BS	Ti100 BM-BS	Ti150 BS	Ti150 BM-BS	Ti200 BS	Ti200 BM-BS
Density (ρ, kg m <sup>-3</sup> )	809.8	812.7	813.1	814.2	814.5	815.4	815.7
Calorific value (CV, MJ kg <sup>-1</sup> )	44.542	45.041	45.275	45.510	45.697	45.921	46.089



**Fig. 3.** FESEM images of soot and other PM emissions collected during the flame combustion of the following fuel samples: (a) Neat diesel, (b) Ti100 BM-BS nanofuel, and (c) Ti200 BM-BS nanofuel.

Table 1. Properties of	f the test	fuel samples
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Ti K

1.6



compustion							
Flomont	Diesel		Ti100 BM-BS		Ti200 BM-BS		_
Element	Weight %	Atomic %	Weight %	Atomic %	Weight %	Atomic %	
СК	65.6	71.7	91.1	94.0	60.4	69.1	
ОК	34.4	28.3	7.1	5.5	34.2	29.3	

0.5

5.4

1.8

**Table 2.** Energy-dispersive X-ray spectroscopy (EDS) analysis of PM emissions during flame combustion.

 $5.55 \times 10^3$  per cm<sup>3</sup>. Fig. 4 represents the normalized number concentration (dN/dlogD<sub>m</sub>) of PM emissions (on a logarithmic scale) at various mobility diameters (D<sub>m</sub>) for titania-nanofuel samples (Ti100 BS, Ti100 BM, Ti100 BM-BS, Ti150 BS, Ti150 BM, Ti150 BM-BS, Ti200 BS, Ti200 BM, and Ti200 BM-BS) in contrast to diesel. The PM emissions during the combustion of diesel majorly consist of elemental carbon (i.e., soot) and organic carbon (unburnt fuel, PAH, etc.). The addition of metal oxide additives promotes better oxidation of diesel fuel, and nano-additives, due to the high specific surface area, improve the heat transfer characteristics of the fuel. Consequently, TiO<sub>2</sub> nano-additives improve the combustion characteristics of diesel and aid in the reduction of soot particles (primary and aggregated soot structures) and unburnt hydrocarbon emissions (including PAHs). BM helps in the size reduction of dispersed nanoparticles, so ball-milled nanofuel samples show a better abatement of PM emissions than bath-sonicated samples. Additionally, the nanofuel samples prepared using BM and BS together, shown the maximum stability and size reduction of the dispersed TiO<sub>2</sub> nanoparticles (Fig. 2), leading to the highest reduction in PM emissions.

Figs. 4(a)-4(c) show the variation in PM emissions for the nanofuel samples prepared using distinct methods (i.e., BS, BM, BM+BS) when diesel is blended with 100 ppm/150 ppm/200 ppm TiO<sub>2</sub> nano-additives. Figs. 4(d)-4(f) represent the change in the number concentration of PM emissions when the different quantity of  $TiO_2$  nanoparticles (i.e., 100 ppm, 150 ppm, and 200 ppm) is added to diesel for BS/BM/BM+BS nanofuel samples. Overall, BM-BS nanofuel samples show a better reduction in PM emissions than BS- or BM-nanofuel samples, and 200 ppm TiO<sub>2</sub>-nanofuel leads to the maximum reduction in PM emissions. Fig. 5(a) reveals the TNC of PM emissions and the percentage change in TNC when diesel fuel is blended with 100 ppm, 150 ppm, and 200 ppm  $TiO_2$  nano-additives. The results show the reduction in TNC of PM emissions with the use of  $TiO_2$ nano-additives. Moreover, the percentage decrement in TNC depends on the morphology of nano-additives dispersed and the amount of TiO<sub>2</sub> nanoparticles added. The morphology of dispersed nano-additives depends on the preparation method of the nanofuel sample. Fig. 5(b) represents the percentage change in the TNC of PM emissions for the different combinations of nanofuel preparation methods and the amount of TiO<sub>2</sub> nanoparticles added to diesel. The percentage reduction in TNC of the PM emissions increases with the amount of TiO<sub>2</sub> nano-additives for each set of nanofuel samples (BS/BM/BM+BS). However, the percentage decrement in TNC diminishes with the increase in ppm of  $TiO_2$  nano-additives. Similarly, the percentage reduction in TNC is greater for ball-milled nanofuel samples compared to just bath-sonicated nanofuel samples, when the same mass of TiO<sub>2</sub> nano-additive is considered. As discussed in section 3.1, BM aids in the size reduction of TiO<sub>2</sub> nanoparticles dispersed in diesel, and the BM-BS nanofuel sample shows the least D<sub>h,avg</sub>, and maximum stability with days (Fig. 2). Thus, BM-BS nanofuel samples lead to the largest reduction in PM emissions for 100/150/200 ppm TiO<sub>2</sub>-laden diesel fuel. Although there is a reduction in TNC of PM emissions with the addition of TiO<sub>2</sub> nanoparticles for each nanofuel sample (Fig. 5), a similar trend cannot be seen in the case of UFP emissions.

In this study, SFR and AFR are maintained at 3.0 L min<sup>-1</sup> and 0.3 L min<sup>-1</sup>, respectively, which fixed the measurement range of  $D_m$  from 15.7–637.8 nm. Therefore, to study the effect of nano-additives on ultra-fine particulate emissions, a number concentration of particles within specific  $D_m$  ranges were considered. To study the effect of TiO<sub>2</sub> nano-additives on UFPs in detail, the overall size range was divided into four specific ranges:

- 1) Sub-23 nm region: 15.7–23 nm,
- 2) Super ultra-fine region: 23–50 nm,
- 3) Ultra-fine region: 50–100 nm, and
- 4) Sub-micron region: 100–637.8 nm.



**Fig. 4.** Measurement of PM emissions with SMPS during the flame combustion of diesel and titania-nanofuel samples: (a)– (c) 100 ppm, 150 ppm, and 200 ppm nanofuel samples, respectively; (d)–(f) bath-sonicated, ball milled, and ball-milled+bath-sonicated nanofuel samples, respectively.

The importance of detailed investigations of UFPs is well-known, considering their impact on human health and the environment (Cassee *et al.*, 2019). The choice of the sub-23 nm range considers the importance of SPN10 (solid particle number > 10 nm) as compared to SPN23 (solid particle number > 23 nm) that is being implemented in Euro-7, European emission standards for vehicles (Samaras *et al.*, 2020). The TNC of PM emissions based on the size range divisions for different fuel samples is shown in Fig. 6. Initial observations show the reduction in TNC for the complete size range of measurement (15.7–637.8 nm); however, on investigating the TNC for D<sub>m</sub> < 100 nm, the results show an increment in the TNC of UFPs even after the suppression in organic and elemental carbon due to the addition of metal oxides, for each nanofuel sample. This can be possibly attributed to the emission of TiO<sub>2</sub> nanoparticles during flame combustion, leading to the enhancement in TNC of PM emissions in the sub-23 nm, super ultra-fine, and ultra-fine regions.





**Fig. 5.** (a) TNC of PM emissions and the percentage change in TNC for TiO<sub>2</sub>-laden nanofuel when compared to diesel, (b) Percentage change in the TNC of PM emissions for distinct nanofuel samples in contrast to diesel.



**Fig. 6.** TNC of PM emissions during flame combustion of diesel and TiO<sub>2</sub>-laden diesel fuel samples for distinct size ranges: All sizes ( $D_m$  = 15.7–637.8 nm),  $D_m$  = 15.7–23 nm,  $D_m$  = 23–50 nm,  $D_m$  = 50–100, and  $D_m$  = 100–637.8 nm.

Although the primary particle size of the TiO<sub>2</sub> nano-additives is relatively small (APS  $\leq$  25 nm), the size of the aggregated structures of the nanoparticles dispersed in diesel is expected to be significantly larger. Considering this, the size of dispersed nano-additives was modified using bath sonication and ball milling (as discussed in section 3.1). The aggregation of nanoparticles was hindered using Span80 surfactant, and the alteration in size morphology of nanoparticles (or aggregated structures) was achieved using mechanical methods like BM and BS. BM aids in the size modification of the dispersed nanoparticles by grinding them to a lower size, and BS helps in the breakdown of weak aggregates. Thus, for nanofuel samples, despite the suppression in soot and other PM emissions, the evolution of TiO<sub>2</sub> nanoparticles leads to the enhancement in the TNC of UFPs.

Fig. 7 represents the change in the TNC of PM emissions for different combinations of size range and nanofuel samples in contrast to diesel. The analysis shows the reduction in the TNC in the sub-micron region (100–637.8 nm) for all the  $TiO_2$ -nanofuel samples. However, the TNC of UFPs increases with the addition of  $TiO_2$  nano-additives, represented by the reddish-yellowish region in Fig. 7. The blue-shade regions depict the reduction in TNC of PM, manifesting the





**Fig. 7.** Percentage change in the TNC of PM emissions during the flame combustion, for distinct size (D<sub>m</sub>) range and TiO<sub>2</sub>-nanofuel samples in contrast to diesel.

benefits of adding TiO<sub>2</sub> nanoparticles. The percentage increment or decrement in the TNC of PM emissions depends on the size morphology of TiO<sub>2</sub> nanoparticles, the amount of nanoparticles added, and the size range for which the analysis has been made. Thus, for a better analysis of the results, the heat map of the percentage change in TNC has been plotted in Fig. 8 based on two divisions: (I) Figs. 8(a)–8(c), the effect of size modification of nanoparticles for a certain mass of nano-additives, and (II) Figs. 8(d)–8(f), the effect of nano-additives quantity for a specific nanofuel preparation method. In Fig. 8, the red-shade zones represent the percentage enhancement in TNC, and the blue-shade zones represent the percentage reduction in the TNC for nanofuel samples compared to neat diesel.

In Fig. 8(a), the 100 ppm ball-milled sample (Ti100 BM) shows the enhancement in TNC of sub-23 nm (15.7–23 nm) and super ultra-fine (23–50 nm) particles to a greater extent as compared to the 100 ppm bath-sonicated sample (Ti100 BS). Even though the BM aids in the size reduction of dispersed nanoparticles and leads to better suppression of soot and PM emissions, the TiO<sub>2</sub> nanoparticles are themselves evolving during the flame combustion, adding up in the TNC of sub-23 nm and super ultra-fine particles. In the case of the Ti100 BM-BS nanofuel sample, the average size of the dispersed nanoparticles is the smallest (see Fig. 2). Still, there is an enhancement of TNC in the super ultra-fine region and not in sub-23 nm. It does not mean that the nanoparticles are not evolving out in the sub-23 nm region. As nano-additives are beneficial due to high specific surface area, the percentage reduction in soot and other PM emissions occurs to a greater extent for Ti100 BM-BS nanofuel in comparison to Ti100 BS and Ti100 BM nanofuel samples, due to which the cumulative TNC of emissions for sub-23 particles shows optimistic results. Similarly, for the ultra-fine region (50-100 nm) and sub-micron region (100-637.8 nm), the results show the percentage reduction in TNC of PM emissions for 100 ppm TiO<sub>2</sub>-nanofuel samples. However, the number concentration of the evolution of nanoparticles will be less than the decrease in the number concentration of PM emissions with the addition of TiO<sub>2</sub> nano-additives, which leads to the net reduction in the TNC. A similar trend can be seen in the case of 150 ppm and 200 ppm nanofuel samples, i.e., Figs. 8(b) and 8(c). The evolution of TiO<sub>2</sub> nanoparticles during flame combustion leads to the increment in TNC, and the regions/zones where the reduction in the TNC of PM emissions dominates will have a net reduction in the TNC. Moreover, one can find that the BM-BS nanofuel sample showed a better reduction in TNC for distinct size ranges and overall PM emissions.

Figs. 8(d)-8(f) represent the percentage change in the TNC of PM emissions with the addition of 100 ppm/150 ppm/200 ppm nano-additives for BS nanofuel, BM nanofuel, and BM-BS nanofuel samples, respectively. The percentage change in TNC has positive values in sub-23 nm and super ultra-fine range for 100 ppm nanofuel sample, depicting that the number concentration of PM emissions increases even after the suppression of soot and PM emissions. The physics behind the enhancement in TNC can be attributed to the evolution of TiO<sub>2</sub> nanoparticles during the flame combustion experiments. Ti100 BM-BS nanofuel sample has the smallest average sized particles



**Fig. 8.** The percentage change in the TNC of PM emissions with distinct size ranges in comparison to diesel, for the following test fuel samples: (a) 100 ppm TiO<sub>2</sub>-nanofuel, (b) 150 ppm TiO<sub>2</sub>-nanofuel, (c) 200 ppm TiO<sub>2</sub>-nanofuel, (d) BS TiO<sub>2</sub>-nanofuel, (e) BM TiO<sub>2</sub>-nanofuel, and (f) BM-BS TiO<sub>2</sub>-nanofuel.

in the dispersed state, which leads to a maximum reduction in PM emissions. Thus, there is an enhancement in TNC in the super ultra-fine region and a reduction in TNC in the sub-23 nm region. The results also show that the percentage reduction in PM emissions in distinct size ranges increases with the amount of nano-additives. Overall, 200 ppm and BM-BS nanofuel samples showed the highest reduction in TNC of PM emissions. Consequently, the addition of TiO<sub>2</sub> nano-additives to diesel aids in the overall reduction of soot and other PM emissions.

Table 3 compares the current research work with previous studies to underline the significance of modifying the size of the dispersed  $TiO_2$  nanoparticles on nanofuel stability. The prolonged stability of nanofuel suspension demonstrated in this work enhances its practical utility for real-life applications. The advantages of reducing the size of nano-additives on PM emissions can be utilized over a long period. As there is no study specific to diesel/biodiesel flame combustion; therefore, the state-of-the-art comparison for the fine and ultra-fine particulate matter emissions reduction with the addition of  $TiO_2$  nano-additives is not presented here. However, in spite of the variations in the primary size of nanoparticles (APS) and surfactant across studies, the improvement observed in the PM emissions follows a similar trend as that of engine particulate emissions characteristics (Geng *et al.*, 2021).



Authors	Deep Fluid /Fuel	Nano-additives (titania)		- Nanafuel properties	Nanofuel
Authors	Base Fluid/Fuel	Quantity	Size (APS)	- Nationuel preparation	stability
(Liu <i>et al.,</i> 2011)	DI water	20 mg L <sup>-1</sup>	10 nm	Ultra-sonication for 2 hours	4 days (conc.
					ratio 4%)
(Leena and Srinivasan, 2015)	Water	500 ppm	15 nm	Ultra-sonication for 3 hours	6 hours
(Venu <i>et al.,</i> 2019)	Palm oil-Biodiesel	25 ppm	19–24 nm	Ultra-sonication for 10 min	4 days
(D'Silva <i>et al.,</i> 2019)	Diesel	$100 \text{ mg L}^{-1}$	< 50 nm	Probe sonication for 30 min	-
(Bello <i>et al.,</i> 2020)	Diesel	50 mg L <sup>-1</sup>	21 nm	Ultra-sonication for 30 min	2 hours
(Vigneswaran <i>et al.,</i> 2021)	Diesel-Water	30 ppm	$50\pm5~\text{nm}$	Stirring 15 min	3 days
Present work	Diesel	100 ppm	≤25 nm	Ultra-sonication for 1 hour,	15 days
				Ball milling for 32 min	

Table 3.	The stability	of nanofluid	suspension	with the addit	ion of TiO <sub>2</sub>	nanoparticles.
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# **4 CONCLUSION**

Soot and other PM emissions during the flame combustion of diesel blended with  $TiO_2$  nanoadditives have been investigated using SMPS. The study delves into the effect of size modification of dispersed nanoparticles and the amount of  $TiO_2$  nanoparticles on PM emissions. The results show the overall reduction in the number concentration of total PM emissions with an increase in the number concentration of particulates in the specific size ranges. The following are the major findings from the experimental work done to investigate the impact of  $TiO_2$  nanoparticles added to diesel on the emission of PM, including UFPs:

- The DLS analysis of the nanofuel samples shows that ball milling aids in the size reduction of the dispersed nanoparticles, and the D<sub>h,avg</sub> is less for BM nanofuel as compared to BS nanofuel, i.e., D<sub>h,avg</sub> (after 24 hours) is 328.2 nm for Ti100 BM and 465.5 nm for Ti100 BS.
- 2) Ti100 BM-BS nanofuel sample shows the smallest average sized particles (D<sub>h,avg</sub> = 208.6 nm, after 24 hours) and the maximum stability in terms of dispersion even after 15 days of nanofuel sample preparation. Thus, ball milling in combination with bath sonication helps in improving the shelf life of the nanofuel.
- 3) BM is a useful mechanical method that changes the morphology of nanoparticles and aids in improving the combustion phenomenon and reducing PM emissions effectively. There is a reduction of 17.26%, 29.09%, and 39.29% in the TNC for Ti100 BS, Ti150 BS, and Ti200 BS nanofuel samples, respectively, whereas the reduction in the TNC is 21.37%, 37.59%, and 45.08% for Ti100 BM, Ti150 BM, and Ti200 BM nanofuel samples, respectively.
- BM-BS nanofuel samples showed the maximum reduction in the TNC for a certain mass of TiO<sub>2</sub> nanoparticles added (37.70% for Ti100 BM-BS, 48.46% for Ti150 BM-BS, and 53.27% for Ti200 BM-BS).
- 5) Nanofuel samples show a reduction in PM emissions; however, the analysis shows an increment in the number concentration of UFPs (< 100 nm). This indicates the evolution of the TiO<sub>2</sub> nanoparticles during flame combustion. The EDS analysis validates the presence of TiO<sub>2</sub> nano-additives evolution in the plume associated with flame combustion.
- 6) The investigations for 100 ppm- and 150 ppm-nanofuel samples reveal the enhancement in the TNC for the super ultra-fine region (23–50 nm). As BM aids in the size reduction of dispersed nanoparticles, the BM nanofuel samples show more increment in the TNC (100.83% for Ti100 BM, 38.25% for Ti150 BM) as compared to BS nanofuel samples (96.46% for Ti100 BS, 7.51% for Ti150 BS) due to the evolution of nano-additives.
- 7) In spite of the maximum reduction in the size of dispersed nanoparticles, the BM-BS nanofuel sample showed the least enhancement of TNC in the super ultra-fine region (16.73% for Ti100 BM-BS, 4.61% for Ti150 BM-BS) for a certain amount of nanoparticles added.
- 8) The TNC in the sub-23 nm region changes (increases or decreases) depending upon the net effect of the evolution of nanoparticles and the suppression of PM emissions for different nanofuel samples.
- 9) 200 ppm nanofuel samples show a reduction in TNC for almost all size ranges. This does not



invalidate the possibility that nanoparticles are evolving during flame combustion. The cumulative effect of percentage reduction in soot (and other PM emissions) and the evolution of  $TiO_2$  nano-additives shows the net decrement in the TNC.

The experimental investigations reveal that with the addition of TiO<sub>2</sub> nano-additives, there is a reduction in the TNC of total PM emissions during the flame combustion of diesel for the entire size range. However, the detailed analysis of UFPs showed the enhancement in the TNC due to the evolution of TiO<sub>2</sub> nanoparticles. The increment or decrement in the number concentration for distinct size (D<sub>m</sub>) depends on the net effect of reduction in PM emissions and evolution of nanoparticles. UFPs play a crucial role in affecting human health and the environment, so the additional burden of nanoparticles in deteriorating air quality needs insight research. Additionally, the measurement of the number concentration of sub-23 nm and UFPs is essential to make the emission norms more stringent. Thus, the detailed investigation of UFPs (1–100 nm) emissions during combustion is essential as the size of primary soot particles lies in the range of a few nanometers. The present study includes the measurement of PM emissions with the help of SMPS using long-DMA that restricts the size range of measurements to 15.7–637.8 nm. Therefore, further studies can be conducted on measuring sub-23 nm and sub-10 nm particles. Additionally, the detailed study of the soot growth mechanism during combustion may help to understand the change in the number concentration of UFPs and submicron particles.

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### SUPPLEMENTARY MATERIAL

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