

90 This study aims to investigate the characterization concentrations of carbonaceous in PM_{2.5} and
91 estimate the source contributions for carbonaceous aerosols.

92

93 **2 METHODS**

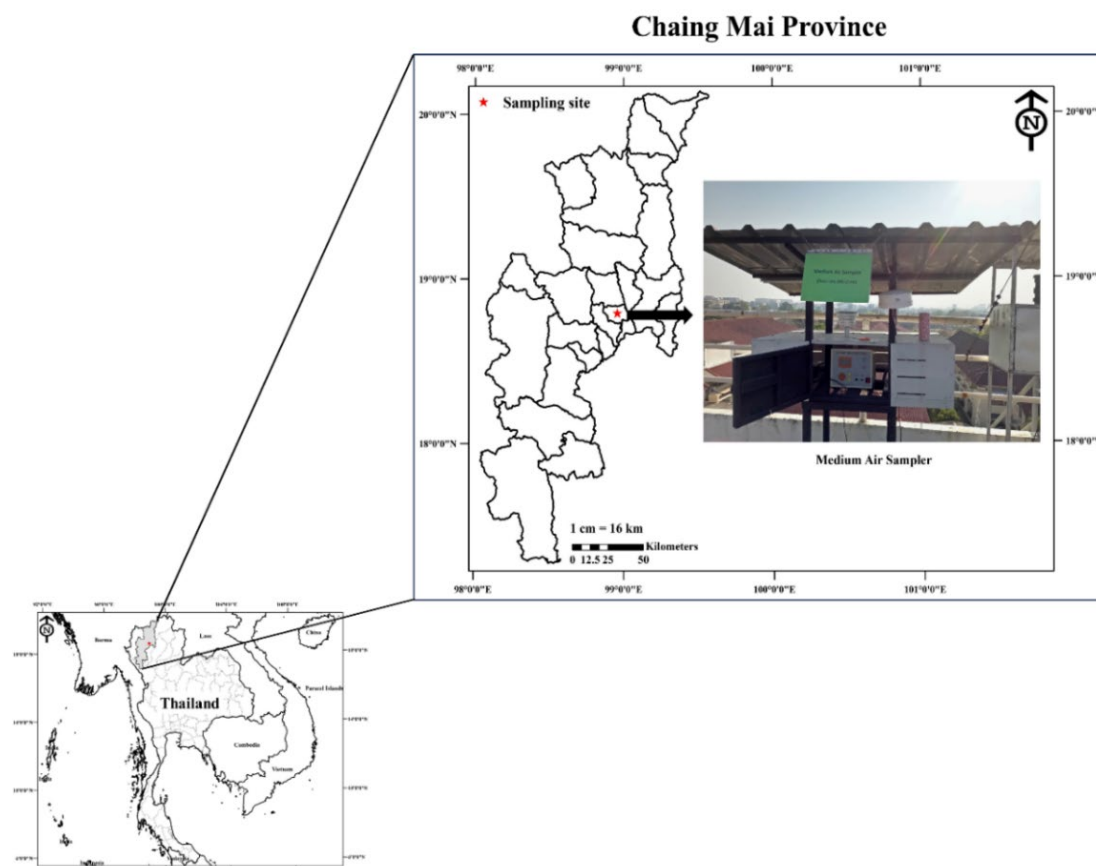
94 **2.1 Sampling site**

95 The study was conducted at the Research Institute for Health Sciences (RIHES) at Chiang Mai
96 University (CMU). Its coordinates are 18° 47' 42.67"N and 98° 57' 28.24"E. PM_{2.5} was collected
97 between January and May 2017 on the rooftop of a four-story building, 30 meters above ground
98 level (Fig. 1). PM_{2.5} samples were collected every two days using quartz filters from Pall Life
99 Sciences (Port Washington, NY, USA). Ambient air data was collected using a 24-hour medium-
100 volume sampler from Qingdao Laoying Ltd (Qingdao, China), with a flow rate set at 100 L min⁻¹.
101 Each quartz filter was heated in a 450°C oven for 6 hours before sampling. A thin aluminum foil
102 covering protected the quartz filters from solar radiation. Before and after sampling, filters were
103 stored in silica gel desiccators for 24 hours. Subsequently, the filters were weighed using a Mettler
104 Toledo microbalance from Switzerland. Following sampling, the filters were placed in a controlled
105 room with a temperature range of 25 ± 2°C and a relative humidity below 40% for measurement.
106 The filters were stored in a freezer at a temperature below -20°C before chemical
107 analysis.

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128
129 **Fig. 1.** The location of the sampling site is in Chiang Mai City. ($18^{\circ} 47' 42.67''\text{N}$ and $98^{\circ} 57'$
130 $28.24''\text{E}$).

131 **2.2 OC, EC and WSOC analysis**

132 The concentration of OC and EC in $\text{PM}_{2.5}$ samples was analysed using a thermal optical
133 transitions OC/EC analyser (Sunset Laboratory, Model-4, U.S.A.) using the thermal-optical
134 transmittance (TOT) method and the NIOSH (National Institute for Occupational Safety and Health)
135 5040 protocol, that improved by Chow et. al. 2007 at the School of Applied Meteorology, Nanjing
136 University of Information Science and Technology (NUIST), Nanjing, China (Chow et al., 2007).

137 In a concise summary, quartz filter disks with a diameter of 17 mm were inserted into a quartz tube
138 within the thermal desorption chamber for analysis. The pyrolysis products performed a redox
139 reaction with manganese dioxide, which caused the conversion of CO₂. The quantification of CO₂
140 was performed with a self-contained nondispersive infrared (NDIR) device. Following the
141 completion of each analysis, a consistent quantity of an internal standard consisting of 5% methane
142 and 95% helium was used, therefore allowing the determination of a predetermined mass of carbon.
143 A weekly calibration was performed using an external sucrose standard at a concentration of 4.21
144 µg L⁻¹ to ensure accurate quantification. A daily calibration was performed using an instrument
145 blank. The detection limits for both OC and EC were 0.5 µg m⁻³.

146 For WSOC analysis, following the method established by Haque et al. (2019), each sample was
147 extracted using a filter with a surface area of 2.26 cm². This quartz filter samples were extracted
148 for 30 min using an ultrasonicator in 10 mL of organic-free ultrapure water, which had a resistivity
149 exceeding 18.2 MΩ cm (measured by a Sartorius Arium 611 UV instrument). Then, filtration using
150 a 0.22 µm PTFE filter to eliminate the insoluble filter matrices. Then, the filtrates obtained went
151 through an analysis for WSOC using a Shimadzu TOC analyzer (model: TOC-Vcsh). The triplicate
152 investigations conducted in the present study revealed that the analytical errors for all carbonaceous
153 components were below 5%, and the reported quantities were within this acceptable range (Haque
154 et al., 2019).

155

156 **2.3 Data analysis**

157 The data obtained from the experiment was collected and evaluated using the SPSS statistical
158 software. The results are presented as the mean value and the standard deviation (S.D.). In addition,
159 the current study employed a Pearson correlation analysis to investigate the relationship between
160 PM_{2.5} concentrations and various chemical compositions of PM_{2.5}.

161

162 **3 RESULTS AND DISCUSSION**

163 **3.1 Temporal Variation of OC and EC**

164 The collection of PM_{2.5} samples was conducted by the Research Institute for Health Sciences at
165 Chiang Mai University (18° 47' 42.67"N and 98° 57' 28.24"E). The sampling continued every two
166 days, covering January 12 to May 30, 2017. The PM_{2.5} concentrations were estimated using the
167 gravimetric method. The range of PM_{2.5} concentrations ranges from 6.07 to 79.9 µg m⁻³, with a
168 mean value of 35.8 ± 16.3 µg m⁻³. The observed mean values were lower than the measurements
169 obtained at the Chiang Mai University campus in 2016 (64.3 ± 17.6 µg m⁻³), which were monitored
170 during the same period (Thepnuan et al., 2019). Additionally, the mean values at the CMU site in
171 2016 were higher than in this study, with a mean of 44.5 ± 32.1 µg m⁻³ (Kawichai, 2021). The mean
172 concentrations of OC and EC were 9.75 ± 4.19 µg m⁻³ and 1.34 ± 0.60 µg m⁻³, respectively, and
173 lower than those from the previous study were 14.6 ± 11.9 µg m⁻³ and 1.80 ± 1.60 µg m⁻³,

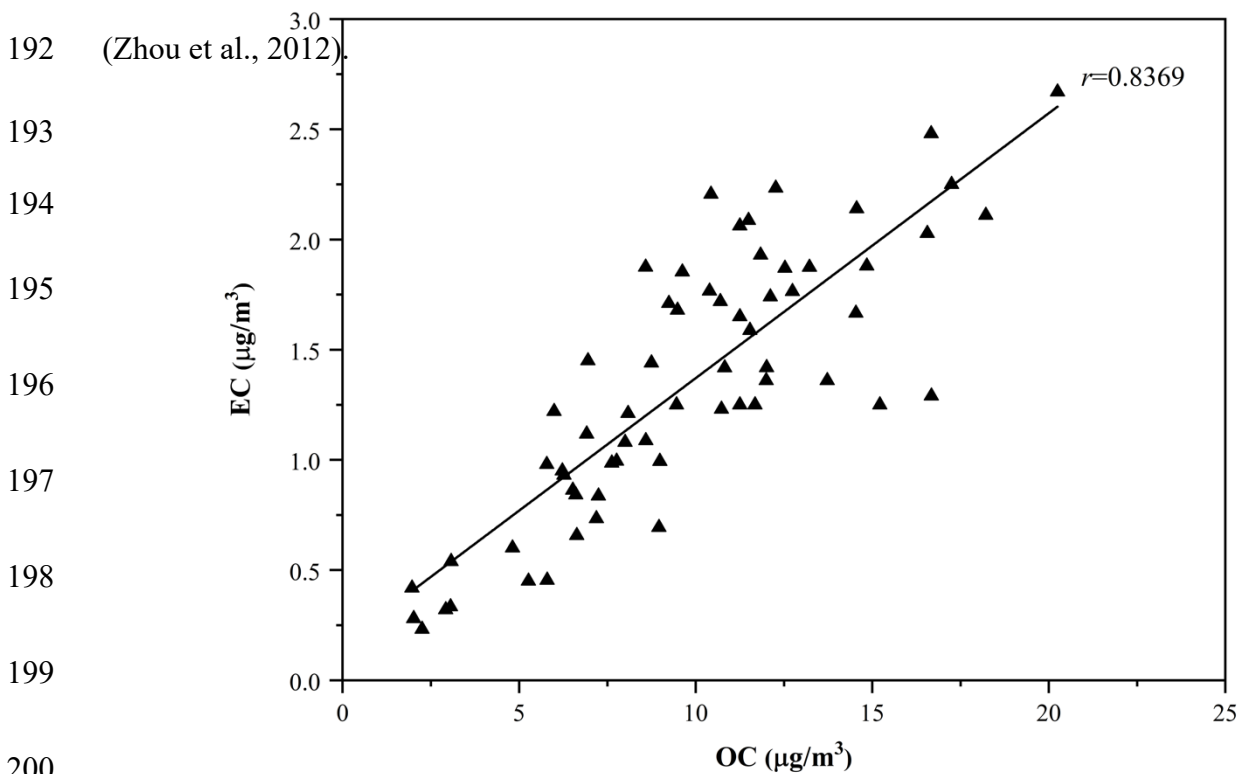
174 respectively (Kawichai, 2021), as shown in **Table 1**. The possible influence of various emission
 175 sources across different seasons might contribute to the higher variability observed in
 176 concentrations of OC and EC. The temporal variations of OC and EC concentrations are mainly
 177 enriched in fine particulates, especially PM_{2.5}. The findings of this study show that the average
 178 concentrations observed were greater compared to the explained levels in urban areas from the
 179 previous studies (Viana et al., 2006; Monteiro dos Santos et al., 2016).

180 **Table 1.** Comparison of OC, EC, OC/EC, WSOC, and WSOC/OC ratio with other studies.

Study site	Sampling period	n	OC ($\mu\text{g m}^{-3}$)	EC ($\mu\text{g m}^{-3}$)	OC/EC ($\mu\text{g m}^{-3}$)	WSOC ($\mu\text{g m}^{-3}$)	WSOC/OC	Reference
Chiang Mai, Thailand	Jan-May 2017	63	9.75 ± 4.19	1.34 ± 0.60	7.65 ± 2.03	7.61 ± 4.08	0.75 ± 0.15	This study
Guangzhou, China	Aug 2006-Aug 2007	52	7.10 ± 3.30	4.00 ± 2.50	no data	2.00 ± 1.20	0.28	(Huang et al., 2012)
Chiang Mai, Thailand	Feb-Aug 2016	30	14.6 ± 11.9	1.80 ± 1.60	15.8 ± 8.01	no data	no data	(Kawichai, 2021)
Chiang Mai, Thailand	Feb-March 2017	34	12.3 ± 4.68	3.01 ± 1.08	4.19 ± 1.10	no data	no data	(Kraisititikul et al., 2024)
Chiang Mai, Thailand	Feb-March 2018	30	10.2 ± 4.10	1.20 ± 0.60	9.40 ± 1.50	8.00 ± 3.40	0.78 ± 0.17	(Song et al., 2022)
Chiang Mai, Thailand	Feb-March 2017	34	12.3 ± 4.68	3.01 ± 1.08	4.19 ± 1.10	no data	no data	(Kraisititikul et al., 2024)
Beijing, China	Spring, 2017	276	9.54 ± 1.08	2.63 ± 1.71	4.38 ± 1.32	5.14 ± 3.25	0.49 ± 0.12	(Jin et al., 2020b)
Tsukuba, Japan	Aug 2017-Jul 2019	62	3.20 ± 1.00	1.20 ± 0.40	3.80 ± 1.40	1.20 ± 0.40	0.40 ± 0.10	(Suto and Kawashima, 2021)

181 The sources of OC are not mainly identified with POC, which is directly emitted into PM_{2.5} from
 182 various sources of emissions such as vehicle exhaust, fuel combustion, and cooking. OC sources
 183 also include SOC, which is formed through photochemical reactions with VOCs and nitrogen
 184 oxides (NO_x). The process of incomplete combustion of biomass burning, coal, or oil from EC. It
 185 can only be detected in primary organic aerosol (Shi et al., 2016). Therefore, it was commonly
 186 employed to monitor primary organic aerosols. These results can potentially provide information
 187 regarding the source of carbonaceous particles. A strong positive correlation ($r = 0.8371, p < 0.001$)

188 was found between OC and EC, as shown in Fig. 2, suggesting the presence of comparable emission
189 sources such as road traffic and biomass-burning (Pant et al., 2015). The correlation between OC
190 and EC implied that there are common sources of emissions for both OC and EC, which may
191 provide helpful information regarding the source of carbonaceous materials in the described region



202 **Fig. 2.** Scatter plot of OC and EC in PM_{2.5} in Chiang Mai City, 2017.

203 The findings from the linear regression analysis conducted on the mass concentration of OC and
204 EC in Chiang Mai City demonstrate a strong association. This suggests that the sources throughout
205 Chiang Mai City display a significant level of similarity (Qi et al., 2018). Furthermore, there is also
206 a strong positive association between PM_{2.5} and TC ($r = 0.8953, p < 0.01$) as well as OC ($r = 0.8969,$

207 $p < 0.01$) (Fig. 3). Additionally, a moderate positive correlation is observed between $PM_{2.5}$ and EC
208 ($r = 0.7558, p < 0.01$). The $PM_{2.5}$ samples had a TC concentration of 32.6%. According to Qi et al.
209 (2018), the findings revealed that compared to Beijing and Tianjin, the $PM_{2.5}$ levels were 31% and
210 25% respectively. Nevertheless, there occurred asynchronous variation in the levels of $PM_{2.5}$
211 concentrations and carbonaceous species. The observed phenomenon can be attributed to the
212 proportion of carbonaceous aerosol, known as total carbonaceous aerosol (TCA), within the $PM_{2.5}$.
213 The present study considers the TCA concentration in the atmosphere, comprised of both EC and
214 organic matter (OM). TCA estimation involves multiplying the OC quantity by a factor of 1.6
215 (Turpin and Huntzicker, 1995). The study conducted in Chiang Mai City found that TCA accounted
216 for 49.8% of the $PM_{2.5}$ mass loading, consequently confirming it as the dominant component of
217 $PM_{2.5}$ (Cao et al., 2005). The results of the present study revealed that the levels of TCA in $PM_{2.5}$
218 were relatively high, with an average of 49.8 ± 13.7 . This observation could perhaps be attributed
219 to the presence of biomass-burning emissions in the immediate area of the sampling site, as shown
220 in previous research (Shi et al., 2016). In previous studies, it was shown that the ambient $PM_{2.5}$ in
221 Hong Kong consists of 17.8% OC and 8.7% EC (Cao et al., 2003), resulting in a TCA of 36.5%.
222 The result indicates that roughly half of the total TCA concentration is present in the environment
223 near roads. The results from prior studies indicate that TCA contributed to approximately $32.0 \pm$
224 8.3% of $PM_{2.5}$ in several research (Niu et al., 2012).

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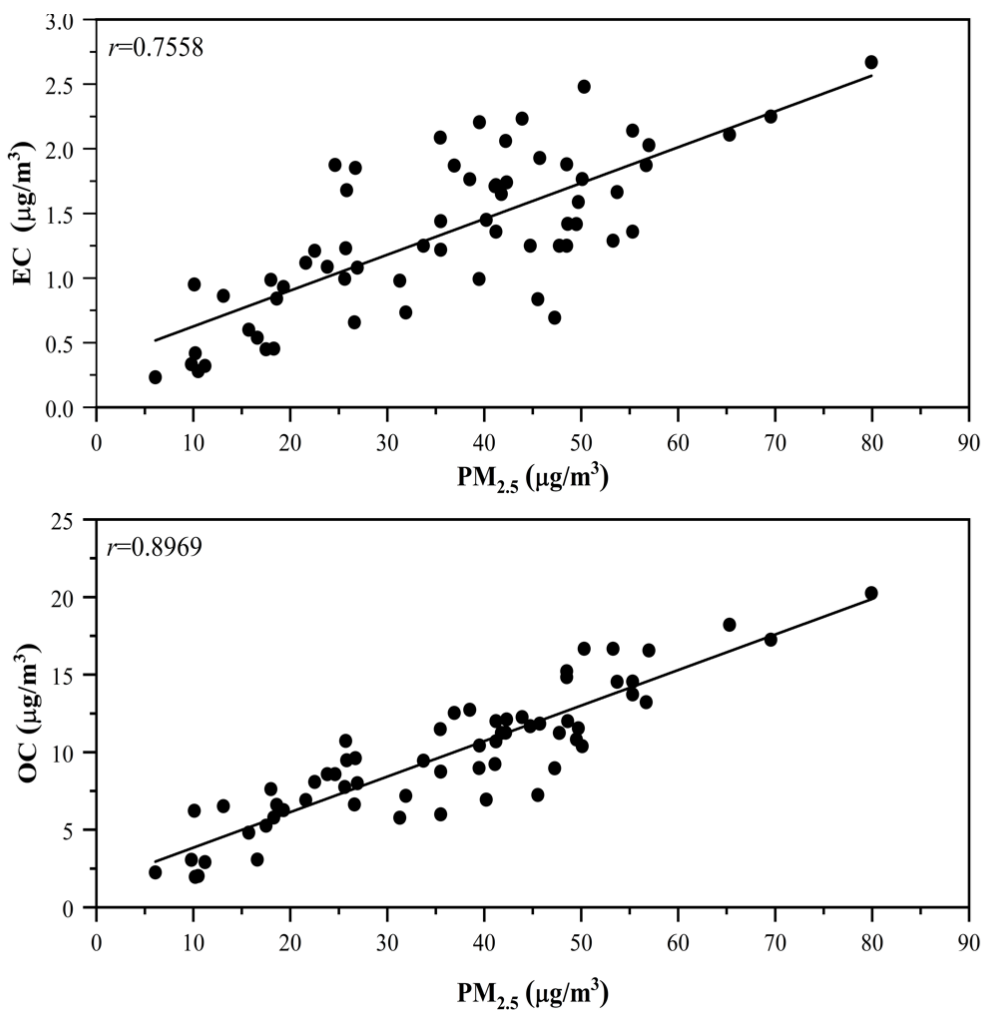


Fig. 3. Correlation between PM_{2.5} and OC, EC in Chiang Mai City, 2017.

241 However, a variation in temporality was observed in the levels of PM_{2.5} concentrations and
 242 carbonaceous species, which can be attributed to the proportion of TCA present in the mass of
 243 PM_{2.5}. The determination of TCA in this study involved the summation of EC and OM. OM was
 244 estimated by multiplying the amount of OC by a factor of 1.6. (Turpin and Huntzicker, 1995). The
 245 present investigation has determined that the average contributions of TCA to the PM_{2.5} mass
 246 amount to 49.8%. The findings of this study confirm the concept that TCA serves as the primary

247 component of PM_{2.5}, according to previous research (Cao et al., 2005). Moreover, the higher
248 fraction of TCA in PM_{2.5} in this region may be attributed to biomass-burning emissions (Shi et al.,
249 2016). Moreover, carbonaceous aerosols have been used to research emission and transformation
250 characteristics. The ratio of OC to EC in PM_{2.5} can be employed to identify the sources of pollution
251 in atmospheric aerosols. The mean ratio of OC to EC in Chiang Mai City was 7.65 ± 2.03 and the
252 period during January and May had the highest OC/EC ratio values (which ranged from 4 to 8),
253 which can be attributed to the prevalence of significant open burning during forest fire incidents
254 within the region. The present study's finding indicate that aerosol carbonaceous particles can
255 potentially contribute to biomass burning (Cao et al., 2005). The observed high OC to EC ratio in
256 this study aligns with findings from other research that have associated it with emissions from wood
257 combustion (Schauer et al., 2002) and biomass burning (Zhang et al., 2007; Samara et al., 2014;
258 Wang et al., 2017). To enhance understanding, an OC/EC ratio over two has been employed as
259 evidence to support the presence of secondary organic aerosols. Schauer et al. (2002) suggest that
260 an OC/EC ratio within the range of 1.0-4.2 indicates emissions originating from diesel and gasoline
261 vehicle exhaust (Schauer et al., 2002). In contrast, a study conducted by Watson et al. 2001 found
262 an OC/EC value of around 9.0, suggesting the significant of biomass-burning activities during the
263 study period (Watson et al., 2001). The OC/EC ratio of the samples that were collected from the
264 CMU site between February 23 and April 28, 2016, was found to be 8.28. The finding given by

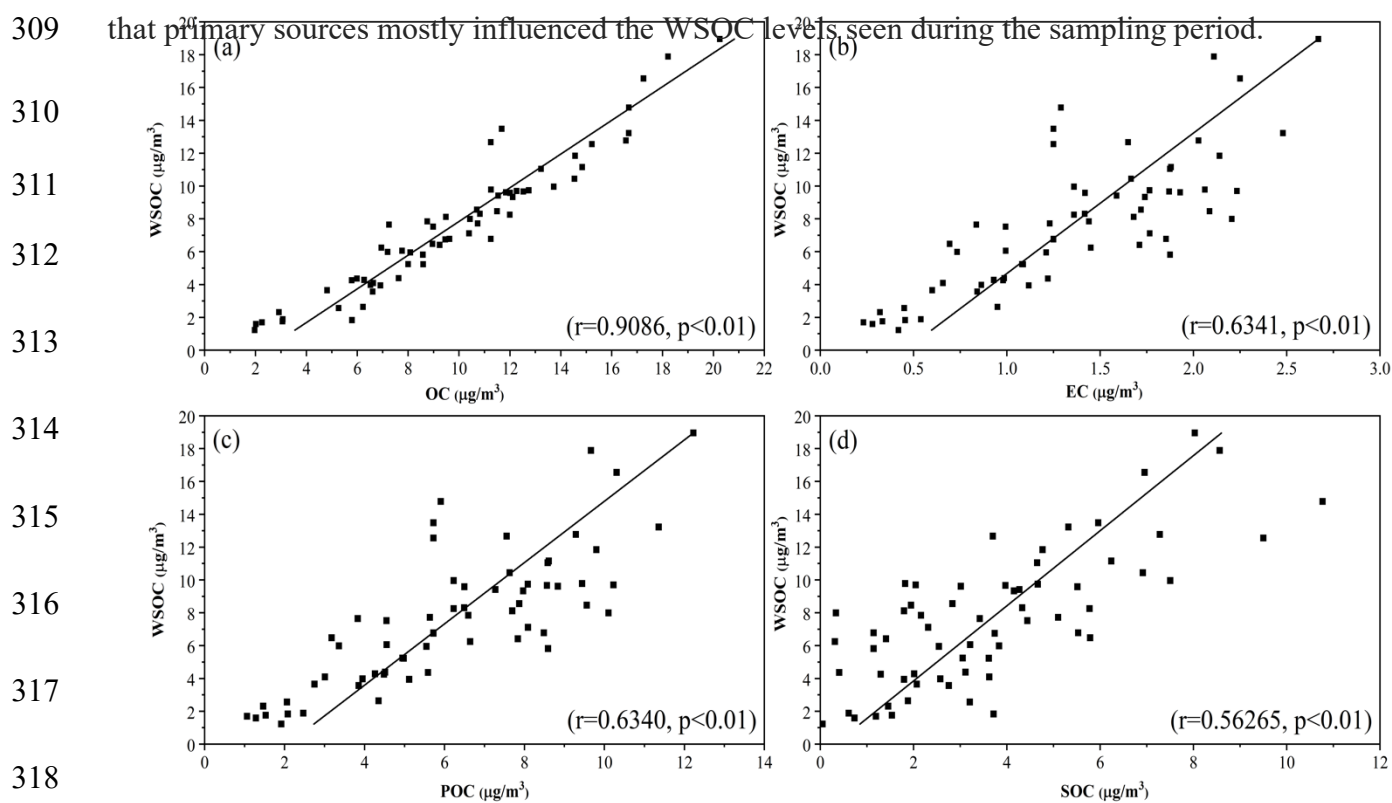
265 Thepnuan et al. (2019) demonstrates an important prevalence of biomass-burning emissions within
266 the study area (Thepnuan et al., 2019). Furthermore, the study found that the average ratios of EC
267 to TC in PM_{2.5} were 0.12 ± 0.03 . Consistent with prior research, EC/TC ratios ranging from 0.1 to
268 0.3 are considered an accurate indicator of a significant presence of carbonaceous substances
269 originating from biomass burning (Engling et al., 2011; Ram and Sarin, 2011; Pani et al., 2019;
270 Kraisitnitikul et al., 2024).

271 **3.2 WSOC Concentration**

272 The study found that the average concentration of WSOC in PM_{2.5} was $7.6 \pm 4.1 \mu\text{g m}^{-3}$, with an
273 observed range of WSOC concentrations over the period was 1.2-18.9 $\mu\text{g m}^{-3}$. The values obtained
274 in the present study were comparable to those found reported for the CMU site in Chiang Mai,
275 Thailand, in 2019, at a similar sampling site. The average concentration recorded was $8.0 \pm 3.4 \mu\text{g}$
276 m^{-3} , while the range of concentrations varied from 1.3-12.5 $\mu\text{g m}^{-3}$ (Song et al., 2022). Moreover,
277 the concentration of WSOC revealed similarities to the levels reported in Urumqi, Northwest China
278 ($6.1 \pm 0.7 \mu\text{g m}^{-3}$). The observed concentration of WSOC was within the range typically associated
279 with emission from biomass burning and the formation of secondary organic aerosol (SOA) (Zhang
280 et al., 2023). The concentration of WSOC in this study was significantly higher when compared to
281 previous studies in Beijing, China ($5.1 \pm 3.2 \mu\text{g m}^{-3}$) (Jin et al., 2020a); Guangzhou, China ($2.1 \pm$
282 $1.7 \mu\text{g m}^{-3}$) (Huang et al., 2012); and Atlanta, U.S.A. ($1.9 \pm 1.0 \mu\text{g m}^{-3}$) (Sullivan et al., 2006). In

283 this study, WSOC accounted for 75% of OC, indicated that the majority of OC in PM_{2.5} in Chiang
284 Mai City is water soluble. Previous research suggests that vehicle emissions tend to have lower
285 levels of WSOC and OC compared to biomass burning sources. This is primarily due to the reduced
286 solubility of the organic components produced during the combustion of liquid fuels such as diesel
287 and gasoline (Ram et al., 2012; Daellenbach et al., 2016). According to the study results of Cheung
288 et al. (2009), the ratios of water-soluble organic carbon (WSOC) to organic carbon (OC) in liquid
289 fuel particles emitted by light-duty vehicles range from 0.06 to 0.19 (Cheung et al., 2009).
290 Saarikoski et al. (2008) reported WSOC/OC ratios of 0.41 and 0.27 for biomass burning and
291 vehicular emissions, respectively (Saarikoski et al., 2008). In an additional research investigation
292 by Daellenbach et al. (2016), WSOC/OC ratios of 0.1, 0.7, and 0.9 were observed for traffic,
293 biomass burning, and oxidized organic aerosol, respectively (Daellenbach et al., 2016). This study
294 determined the mean WSOC/OC ratio in PM_{2.5} in CMU was 0.75 ± 0.02 . The ratio employed as an
295 indicator of the aging process of aerosol (Cho and Park, 2013) and the highest WSOC/OC ratio
296 observed in this study indicate that the OC was more oxidized in the atmosphere during the
297 sampling period (Huang et al., 2012). The WSOC/OC ratios can become a reliable measure for the
298 occurrence of SOA due to the significant solubility of the majority of SOA in water (Kondo et al.,
299 2007; Weber et al., 2007; Huang et al., 2012). Additionally, an analysis was conducted to calculate
300 the correlation coefficient between WSOC and other species in the samples collected at the CMU

301 site. Fig. 4. illustrates a positive correlation between the WSOC and OC with a correlation
 302 coefficient of 0.9086 ($p < 0.01$). This finding suggests that the presence of OC and primary
 303 combustion emissions considerably influenced the abundance of WSOC during the study period.
 304 The study conducted by Huang et al. (2012) found a modest correlation between WSOC and EC,
 305 likely due to the fuel combustion and biomass burning being the main contributors to EC (Huang
 306 et al., 2012). Once again, the findings from the WSOC analysis conducted by WSOC demonstrated
 307 a statistically significant association with POC ($r = 0.6340$, $p < 0.01$) using the EC-tracer method.
 308 This correlation stronger than the correlation with SOC ($r = 0.5625$, $p < 0.01$). These results suggest



320 **Fig. 4.** Correlation of WSOC with (a) OC, (b) EC, (c) POC, and (d) SOC.

321

322 **3.3 Estimation of secondary organic aerosol concentrations**

323 This study shows that the measurement of EC may be a useful measure for POC and SOC. EC
324 comprises non-combustion primary organic carbon and primary organic carbon formed through
325 combustion processes. The recommended method for EC tracing employed in this work relates to
326 combustion sources, including primary organic components. In addition, there are many methods
327 for measuring TOC; the concentration of SOC could be calculated by the experiential equation,
328 following Castro's equation (Turpin and Huntzicker, 1995).

$$329 \quad \text{POC} = \text{EC} \times (\text{OC/EC})_{\text{primary}} \quad (1)$$

$$330 \quad \text{SOC} = \text{TOC} - \text{POC} \quad (2)$$

331 where "OC/EC primary" refers to an estimate of the minimum ratio between OC and EC.
332 Nevertheless, the primary ratio of organic carbon to elemental carbon can be occasionally
333 unavailable due to various factors, including the emission type, ambient temperature, and method
334 applied to carbon determination. The minimum observed ratios of organic carbon to elemental
335 carbon (OC/EC) in this study were found to be 4.58, with the highest ratio being observed in April.
336 According to Feng et al., 2009, the mean concentration of estimated SOC in PM_{2.5} was found to be
337 $2.68 \pm 2.38 \mu\text{g m}^{-3}$. This value was lower than the annual average concentrations of estimated SOC
338 at ZB and JD sites in Shanghai, which were reported as 7.2 and 5.3 $\mu\text{g m}^{-3}$, respectively (Feng et
339 al., 2009). Furthermore, the occurrence of photochemical reactions in the area might account for

340 the lowest OC/EC ratios, suggesting that incomplete combustion emissions from sources such as
341 motor vehicle exhaust, fuel burning, and biomass burning play a significant role in the overall OC
342 mass in Chiang Mai City (Wu et al., 2015; Kaskaoutis et al., 2020).

343 Based on the findings, it was observed that the SOC/OC ratio in Chiang Mai City was 36.7%
344 higher than compared to the values obtained from the CMB modeling results in urban areas, which
345 ranged between 18-19% (Feng et al., 2006). The results suggest that SOC was the primary
346 constituent of OC, and that SOC played a significant part in forming atmospheric particles within
347 Chiang Mai City. The observed pattern displayed a difference when compared to the findings of
348 Niu et al., 2012, and Qi et al. 2018, since their studies have shown higher values than the results
349 from our studies (Niu et al., 2012; Qi et al., 2018). On the other hand, urban areas defined by fossil
350 fuel combustion revealed the lowest SOC/OC ratio, suggesting that carbonaceous species primarily
351 originated from motor vehicles and other local pollutants, such as transportation (Gu, 2010). During
352 the study period, the mean concentration of TOC released by POC was $6.15 \mu\text{g m}^{-3}$. The POC to
353 OC ratio for January to May has been reported as 63.0, 74.4, 62.4, 54.0, and 63.3%, respectively.
354 The study's results showed that biomass combustion was the predominant contributor to the source
355 of OC during this period (Ram et al., 2008; Kaskaoutis et al., 2020). Biomass burning usually
356 provides significant amounts of organic matter from plant spores, pollen, and non-combustion
357 emissions from industrial sources (Ding et al., 2008; Klejnowski et al., 2017; Hong et al., 2019).

358 Nevertheless, it was observed that the concentration of SOC showed more of an increase compared
359 to POC during the season. During this region's dry season, it could be possible that increased levels
360 of photochemical oxidants, such as volatile organic carbon emissions, can be attributed to the
361 expected increased traffic in metropolitan regions and the release of biogenic compounds from
362 plants. The results of this study demonstrated a slight correlation between SOC/OC and PM_{2.5},
363 suggesting that the formation of secondary species in the Pear Review Delta is mainly influenced
364 by gas-phase photochemical processes occurring with particle phase (Poppe et al., 1993; Kang et
365 al., 2004). In addition, it is essential to note that secondary organic aerosol (SOA) constitutes a
366 significant amount of organic aerosol, resulting from the chemical transformations of VOCs
367 through both homogeneous and heterogeneous processes (Claeys et al., 2004).

368 The determination of the concentration of SOA was achieved by multiplying the concentration
369 of SOC by a factor of 1.6 (He et al., 2015). The study determined that the mean concentration of
370 SOA in Chiang Mai City was $5.89 \pm 3.80 \mu\text{g m}^{-3}$, accounting for roughly 17.0% of PM_{2.5}. This
371 value was lower than the reported concentrations in Taiyuan and at the JD site in Shanghai City
372 (He et al., 2015; Feng et al., 2019). The research findings revealed that SOA emission at the study
373 site were relatively low, contributing only a small portion to the overall concentration of PM_{2.5}. A
374 temporal analysis of SOA concentration demonstrated elevated levels during March and April,
375 particularly during periods marked by high levels of smoke haze in the area. Furthermore, it has

376 been shown that the SOA does not contribute to a significant proportion of PM_{2.5}. This observation
377 illustrates the significant impact of wildfire smoke, demonstrating its crucial role in influencing
378 the atmospheric composition in this area (Keywood et al., 2011).

379

380 **4 CONCLUSIONS**

381 The present study investigated the characteristics of PM_{2.5}, OC, and EC in Chiang Mai City,
382 Thailand, specifically during the smoke haze episode. The research revealed that biomass burning
383 was identified as the primary contributor of OC in PM_{2.5} during the period of smoke haze in Chiang
384 Mai City, Thailand. This was supported by the significantly high percentage of POC relative to
385 TOC. Additionally, the high ratio of OC/EC observed in PM_{2.5} was attributed to an enhanced
386 incidence of biomass burning. The combustion of biomass is widely recognized for being able to
387 release significant amounts of primary organic matter. The highest value of the WSOC/OC ratio
388 was seen during the sample period, providing significant evidence to support the assumption that
389 WSOC is a significant constituent of OC. This observation implies that the primary sources that
390 contribute to WSOC also have the possibility of being the primary sources of OC. Moreover, the
391 research also revealed that contributions from SOC, accounting for 39.5% of TOC, indicate a
392 higher presence of air pollution and its influence on driving the condensation or adsorption of
393 volatile organic compounds. A higher level of photooxidation of precursors was seen during the
394 dry season due to high ambient temperatures.

395 **ACKNOWLEDGMENTS**

396 The authors are grateful for the research funding provided by the CMU Junior Research
397 Fellowship Program (to S.K.). The authors also thank the Research Institute for Health Sciences at
398 Chiang Mai University in Chiang Mai, Thailand, for laboratory and field research support. The
399 study was performed in collaboration with the School of Applied Meteorology at Nanjing
400 University of Information Science and Technology in Nanjing, China.

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